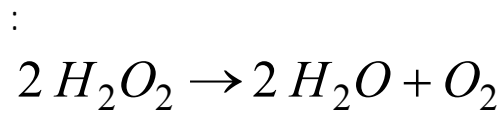


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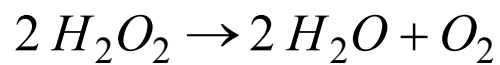
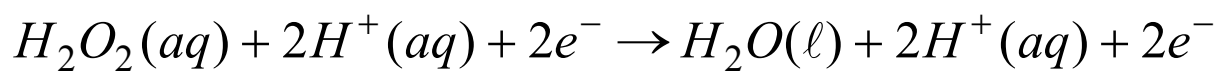
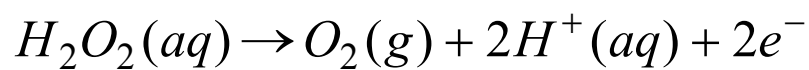
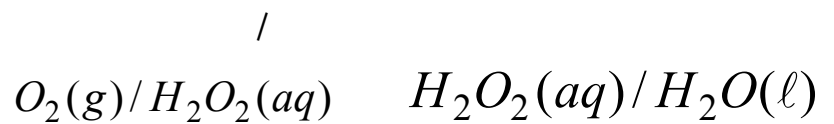
1 litre

x (vol)

$H_2O_2(aq)$

x litre

(P = 101,3 Kpa 0°C)



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50ml (I)

$C = 1.10^{-2} \text{ mol/l}$

$H_2C_2O_4(aq)$

$T = 25^\circ C$

50ml (II)

$C = 1 \text{ mol/l}$

$H_2C_2O_4(aq)$

$T = 25^\circ C$

50ml

$H^+(aq)$

$K^+(aq) + MnO_4^-(aq)$

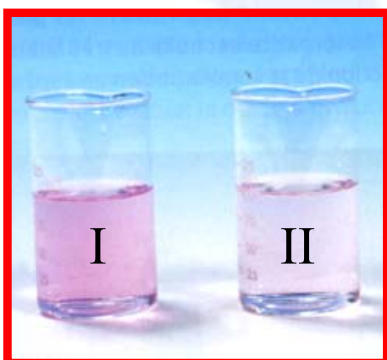
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-2-

(II)

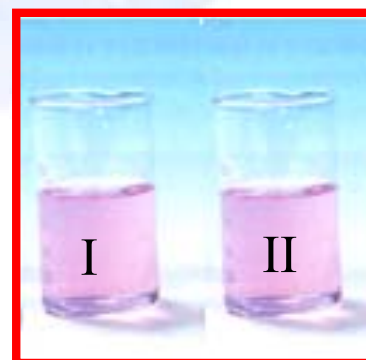
(I)

-1-



شكل - 2 -

5 دقائق بعد المزج



شكل - 1 -

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(II)

- 1

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- 2

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50ml (I)

-

$$C = 1.10^{-2} \text{ mol/l}$$

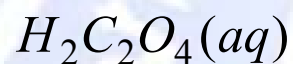


$$T = 25^\circ C$$

50ml (II)

-

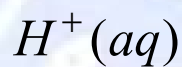
$$C = 1.10^{-2} \text{ mol/l}$$



$$T = 45^\circ C$$

50ml

-



5

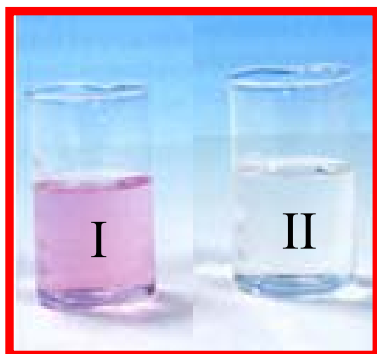
-2-

(II)

(I)

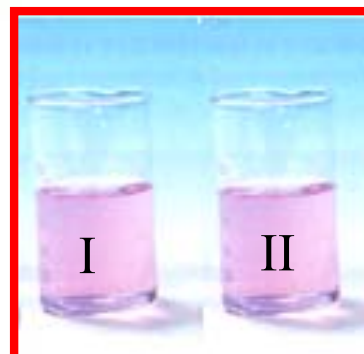
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شكل - 2 -

5 دقائق بعد المزج



شكل - 1 -

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(II)

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- 2

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$t = 0$

(modification permanente)

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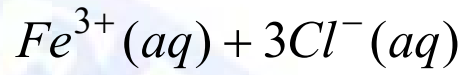
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(III)

$$V = 60 \text{ mL}$$

$$C = 1 \text{ mol/l}$$



$$C_2 = 0,5 \text{ mol/l}$$

$$C_1 = 1 \text{ mol/l}$$

$$C_3 = 0,1 \text{ mol/l}$$

. (20 Vol)

50 mL

1 mL

- 1

(III)

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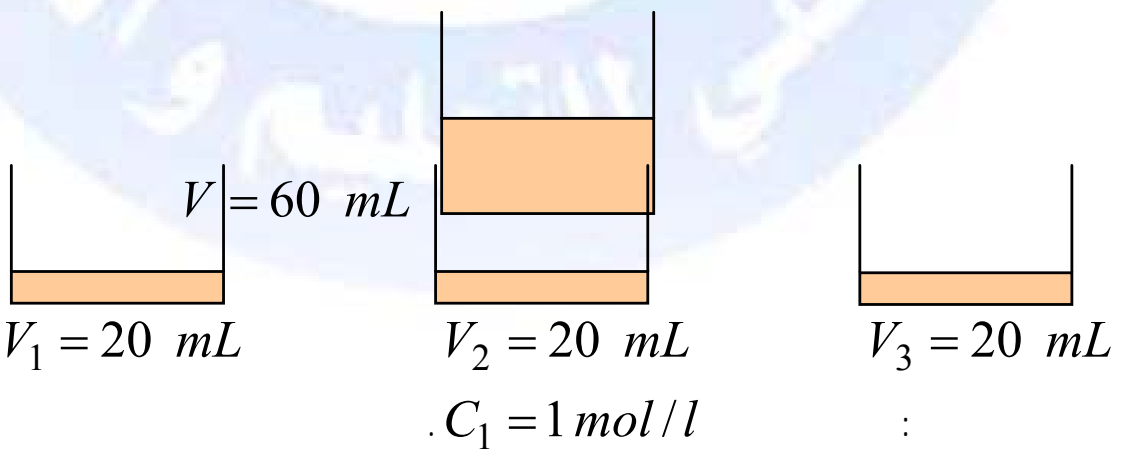
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$$V_1 = V'_1 = 20 \text{ mL} , C_1 = 1 \text{ mol/l}$$

$$. C_2 = 0,5 \text{ mol/l} :$$

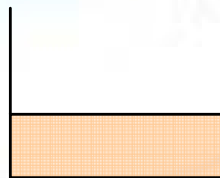
$$n = \frac{C_1}{C_2} = \frac{1}{0,5} = 2 :$$

$$n = \frac{V'_2}{V_2} = 2 :$$

V'_2

$$V'_2 = 2 \times V_2 = 2 \times 20 = 40 \text{ mL} :$$

20mL :



$$V'_2 = 40 \text{ mL} , C_2 = 0,5 \text{ mol/l}$$

$$. C_3 = 0,1 \text{ mol/l} :$$

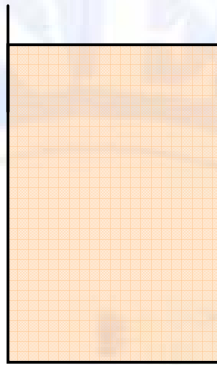
$$n = \frac{C_1}{C_3} = \frac{1}{0,1} = 10 :$$

$$n = \frac{V'_3}{V_3} = 10 :$$

V'_3

$$V_3' = 10 \times V_3 = 10 \times 20 = 200 \text{ mL} :$$

$$180 \text{ mL} :$$



$$V_3 = 200 \text{ mL} , C_3 = 0,1 \text{ mol/l}$$

(III)

– 2

– 3

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– 2

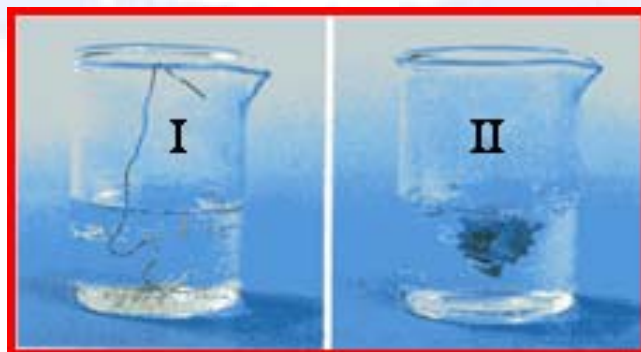
(20 vol)

50ml (I)

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$T = 50^\circ\text{C}$

$\text{Pt}(s)$



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- 3

20mL

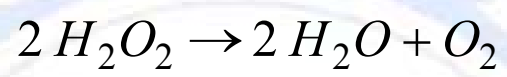


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